

# The Modern Role of NDDO-Based Semiempirical MO Theory

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NDDO-based MO theory has some characteristic strengths that define its role in modern research. These are primarily its applicability to large systems (100,000 atoms without local approximations<sup>1</sup> or 50,000 atom repeat-units in periodic calculations<sup>2</sup>), its good performance for one-electron properties<sup>3</sup> and excited states.<sup>4</sup> However, today's most widely used Hamiltonians suffer from a plethora of non-orthogonal two-center correction terms that improve the performance for weak interactions,<sup>5</sup> or even specific types of bond, such as C-C triple bonds.<sup>6</sup>

This lecture will describe software for very large-scale NDDO calculations and give examples of studies of charge transport in molecular electronic devices and biological systems.

The problem of the “misuse” of nucleus-nucleus potentials will be discussed and a “Feynman” dispersion correction introduced that does not depend on a modified internuclear potential.

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<sup>1</sup> M. Hennemann and T. Clark, *J. Mol. Model.* **2014**, *20*, 2331.

<sup>2</sup> J. T. Margraf, M. Hennemann, B. Meyer and T. Clark, *J. Mol. Model.* **2015**, *21*, 144.

<sup>3</sup> H. B. Thomas, M. Hennemann, P. Kibies, F. Hoffgaard, S. Güssregen, G. Hessler, S. M. Kast and T. Clark, *J. Chem. Inf. Model.* **2017**, *57*, 1907–1922.

<sup>4</sup> M. R. Silva-Junior and W. Thiel, *J. Chem. Theory Comput.* **2010**, *6*, 1546–1564; P. O. Dral and T. Clark, *J. Phys. Chem. A*, **2011**, *115*, 11303–11312.

<sup>5</sup> J. Hostaš, J. Rěžáč and P. Hobza, *Chem. Phys. Lett.* **2013**, *568*, 161–166.

<sup>6</sup> J. J. P. Stewart, *J. Mol. Model.* **2007**, *13*, 1173–1213.