

Theoretical Study of Clusters of HNO₃, H₂O₂, SO₂ and H₂O.

Kenneth Elkner,¹ R. C. Binning, Jr.,² Daniel E. Bacelo^{2,3}

¹*Cupeyville School, Cupey, Puerto Rico 00928*

²*Department of Sciences and Technology, Universidad Metropolitana, P. O. Box 21150, San Juan, Puerto Rico 00928-1150, USA.*

³*Dpto. de Química, FCN, Universidad Nacional de la Patagonia San Juan Bosco, Km. 4, (9000) Comodoro Rivadavia, Chubut, Argentina.*

Abstract

Clusters of HNO₃, H₂O₂ and SO₂, important atmospheric trace species, have been examined in DFT optimizations with the BPW91 functional and numerical basis sets. In addition several structures of the three subject molecules clustered with as many as 16 water molecules have been optimized to examine general aspects of stability. HNO₃, H₂O₂ and SO₂ form a number of stable clusters. The most stable water clusters were those in which the water molecules cohere as they would essentially in the absence of solutes and the solute molecules reside on the surface. The results indicate that reactions among the solute molecules in atmospheric clusters are most likely surface reactions even at elevated temperatures.